

Ag(I)-Catalyzed Three-Component Reaction of 2-Alkynylbenzaldehydes, Amines, and Diazo Compounds

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(5) Supporting Information

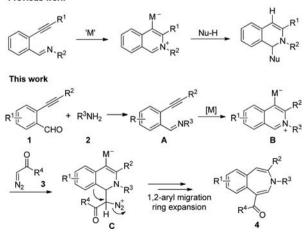


ABSTRACT: Diazo compounds have been employed as the nucleophile in a silver-catalyzed three-component reaction with amines and 2-alkynylbenzaldehydes. Various 3-benzazepines were prepared in a one-pot manner based on a cascade imine—-yne cyclization/nucleophilic addition/1,2-aryl migration process. Moreover, this Ag(I)-mediated reaction also provides a practical route to diazo-containing dihydroisoquinolines under slightly modified conditions.

M ulticomponent reactions are among the most efficient synthetic methods, and they allow the construction of complex molecular architectures from simple precursors with high levels of stereocontrol in a single operation.¹ In the past decades, transition-metal-catalyzed tandem reactions of 2-alkynylbenzaldehydes, amines, and nucleophiles have been developed as efficient methods for the generation of 1,2-dihydroisoquinoline derivatives (Scheme 1).² A variety of different nucleophiles, such as terminal alkynes,³ organic metal reagents,⁴ ketones,⁵ indoles,⁶ phosphites,⁷ and TMSCF₃,⁸ have

Scheme 1. Three-Component Reaction of 2-Alkynylbenzaldehydes, Amines, Nucleophiles, and Our Design

Previous work



been used. However, diazo compounds have never been reported as one of the key reactants in the coupling reactions with 2-alkynylbenzaldehydes and amines.⁹ We assumed that the related nucleophilic addition of the diazo compound to isoquinolinium B would form intermediate C,¹⁰ which might afford a 3-benzazepine skeleton via a 1,2-aryl migration process.¹¹ The challenge in the development of an efficient imine-yne cyclization/nucleophilic addition/1,2-aryl migration cascade process lies in the restraint of possible side reactions, such as the formation of β -ketone esters¹² or N–H insertion of amines with diazo compounds.¹³ Moreover, 1,2-H¹⁴ and 1,2-N migration¹⁵ can also compete with 1,2-aryl migration from intermediate C. We envisioned that employing judicious reaction conditions, in particular an appropriate transitionmetal catalyst, would allow the reaction to proceed along the desired pathway. As part our recent interest in the nucleophilic addition of diazo compounds to in situ generated iminium ions,¹⁶ herein we report a silver-catalyzed three-component reaction of 2-alkynylbenzaldehydes, amines, and diazo compounds. This novel reaction provides an efficient and convenient synthetic route to 3-benzazepines, which are a prominent substructure in numerous pharmaceuticals and naturally occurring alkaloids.¹⁷

Toward this objective, we started our optimization by using 2-(phenylethynyl)benzaldehyde 1a (0.2 mmol), aniline 2a (1.05 equiv), and ethyl diazoacetate 3a (1.1 equiv) as the model substrates. Initially, various transition-metal catalysts, such as $Cu(OTf)_2$, $FeCl_3$, $InCl_3$, and $PdCl_2$, were tested. Although

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these metal salts have been reported as good catalysts for imine—yne cyclization, 3a,4b,18 their application in the present three-component reactions resulted in complex mixtures (Table 1, entries 1–4). Gratifyingly, when 10 mol % of AgOTf was

Table 1. Optimization of Reaction Conditions^{*a,b*}

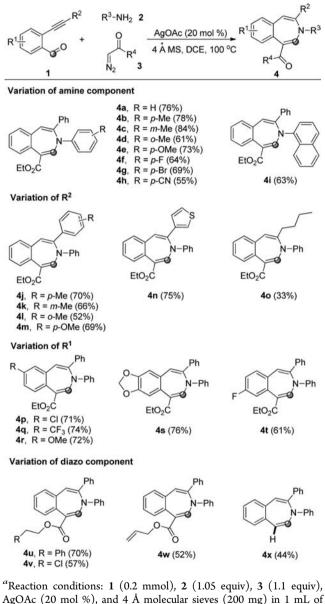
\bigcirc	Ph PhNH ₂ 2a + CO ₂ Et CHO N ₂ 1a 3a	Ag(I) catalyst solvent, temp	EtO ₂ C 4a	Ph +	CO ₂ E1
entry	catalyst (mol %)	temp (°C)	solvent	4a (%)	5a (%)
1	$Cu(OTf)_2$ (10)	75	DCE	9	0
2	$FeCl_3$ (10)	75	DCE	0	7
3	$InCl_3$ (10)	75	DCE	0	13
4	$PdCl_2$ (10)	75	DCE	0	0
5	AgOTf (10)	75	DCE	21	25
6	AgF (10)	75	DCE	15	37
7	AgNO ₃ (10)	75	DCE	19	29
8	Ag_2CO_3 (10)	75	DCE	28	24
9	$AgSbF_6$ (10)	75	DCE	16	28
10	AgOAc (10)	75	DCE	47	23
11	AgOAc (10)	85	DCE	68	15
12	AgOAc (10)	100	DCE	77	9
13	AgOAc (10)	120	DCE	51	5
14	AgOAc (20)	100	DCE	83	<2
15	AgOAc (20)	100	MeCN	21	65
16	AgOAc (20)	100	DMF	17	63
17	AgOAc (20)	100	dioxane	27	51
18	AgOAc (20)	100	toluene	19	58
19	AgOTf (10)	25	DMF	0	91 ^c

^{*a*}Reaction conditions: 2-(phenylethynyl)benzaldehyde 1a (0.2 mmol), aniline 2a (1.05 equiv), ethyl diazoacetate 3a (1.1 equiv), 4 Å molecular sieves (200 mg), and sliver catalyst in 1 mL of solvent for 24 h. ^{*b*}Yields were determined using mesitylene as an internal standard. ^{*c*}Isolated yield.

employed, the desired 3-benzazepine 4a was obtained in 21% yield concomitant with the formation of an isolable diazo product 5a in 25% yield (Table 1, entry 5). Since the reaction was carried out under neutral conditions, the formation of diazo compound 5a must occur through the nucleophilic addition of diazoacetate 3a to isoquinolinium B, followed by deprotonation/protonation from intermediate C. Encouraged by this result, several silver salts, including AgF, AgNO₃, Ag₂CO₃, AgSbF₆, and AgOAc, were examined; among them, AgOAc gave the best yield of 3-benzazepine 4a (Table 1, entries 6–10). We found that the formation of 4a was more favorable at elevated temperatures (Table 1, entries 11 and 12). However, increasing the temperature to 120 °C has a negative impact on the reaction, resulting in the fast decomposition of ethyl diazoacetate 3a (Table 1, entry 13). We found the yield of diazo compound 5a could be further suppressed by using 20 mol % of AgOAc as the catalyst, in which the desired 3benzazepine 4a was obtained in 83% yield (Table 1, entry 14). Varying the solvent from DCE to MeCN, DMF, 1,4-dioxane, or toluene gave low yields of 4a (Table 1, entries 15–18). In these cases, the diazo compound 5a was obtained as the major product. Notably, 91% yield of 5a could be isolated when the reaction was carried out in DMF at room temperature by using 10 mol % of AgOTf as the catalyst (Table 1, entry 19).

Having established the optimized reaction conditions, an investigation into the versatility and functional group tolerance of this reaction process was performed. First, the amine component was varied, and their coupling with 2-(phenyl-ethynyl) benzaldehyde 1a and ethyl diazoacetate 3a were investigated. As shown in Scheme 2, the reaction proceeded

Scheme 2. Scope of the Ag(I)-Catalyzed Three-Component Reactions a,b



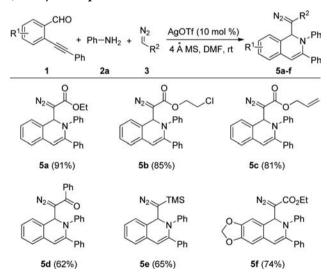
DCE at 100 °C for 24 h. ^bIsolated yields.

smoothly with *o-*, *p-*, and *m*-methyl-substituted anilines as the substrates. The optimal conditions were compatible with a variety of substitutents, including methoxy, fluoro, and cyano groups. It is noteworthy that a bromo substituent could survive in the reaction, which provided the possibility of functionalizing the resulting 3-benzazepines by using palladium-catalyzed coupling reactions. We were delighted to find that 1-aminonaphthalene was also consistent with the optimal conditions, leading to 3-benzazepine **4i** in 63% yield. Unfortunately, when alkylamines such as 1-hexylamine and

benzylamine were employed as the substrates, the reactions gave complex mixtures, and only trace amounts of the desired products were detected. Next, various 2-alkynyl benzaldehydes 4j-t were examined under the optimized reaction conditions. The reaction was not significantly affected by the substitutent on the phenyl ring at the acetylenic center (R^2) ; products 4j-m were isolated in 52-70% yields. Benzaldehyde with triplelinked thiophene also underwent the reaction smoothly with phenylamine and ethyl diazoacetate to assemble the desired 3benzazepine **4n** in 75% yield. Notably, *o*-alkylacetylenicsubstituted benzaldehyde was also suitable substrate for the reaction but afforded the 4-butyl-3-benzazepine 40 in low vield. When a substitutent was introduced to the para or meta position of the aldehyde, the three-component coupling products 4p-s were obtained in good yields. Both electronwithdrawing groups, such as Cl, F, and CF₃, and electrondonating groups, such as OMe and OCH2O, could be well tolerated in this transformation. Finally, we tested the reactions with different α -diazo esters, and they all underwent smooth reactions with 1a and 2a to give the corresponding 3benzazepines 4u, 4v, and 4w in yields of 70%, 57%, and 52%, respectively. In the case of TMSCHN₂, product 4x with deprotection of TMS group was isolated as the major product.

As shown in Table 1, the reaction could afford an isolable diazo-containing dihydroisoquinoline 5a in an excellent yield by varying the solvent and temperature (entry 19). The scope of the present Ag(I)-mediated three-component reaction for the preparation of these diazo-containing cycloadducts was also examined (Scheme 3). To our delight, the reaction proved to

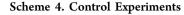
Scheme 3. Sliver-Catalyzed Synthesis of Diazo-Containing 1,2-Dihydroisoquinolines^{a,b}

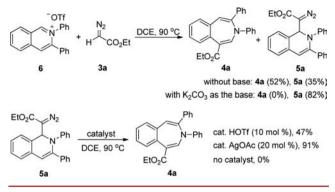


^{*a*}Reaction conditions: benzaldehyde 1 (0.2 mmol), aniline 2 (1.05 equiv), diazo compound 3 (1.1 equiv), AgOTf (10 mol %), and 4 Å molecular sieves (200 mg) in 1 mL of DMF at rt for 12 h. ^{*b*}Isolated yields.

be quite general with various diazo compounds, including α diazo esters, α -diazo ketone, and TMSCHN₂ as the substrates. In all cases, the desired β -amino- α -diazo compounds **5a**-**f** were isolated in good yields, and no ring-expansion products were detected.

To understand the mechanism of the reaction, several control experiments were performed (Scheme 4). First, heating

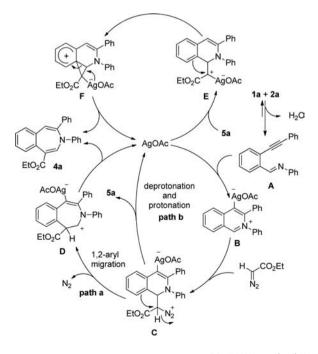




the solution of diazoacetate 3a and isoquinolinium salt 6 gave a mixture of 3-benzazepine 4a and diazo compound 5a in a ratio of 1.5:1. When 1 equiv of K₂CO₃ was added, the formation of 4a was completely suppressed, and compound 5a was isolated as the sole product in 82% yield. These results indicated K₂CO₃ might promote the deprotonation to form 5a. Since the reaction of 6 and 3a generates HOTf, which can be used as a catalyst for 1,2-aryl migration,¹⁹ the addition of a base also neutralized the HOTf in the solution. To evaluate the catalytic efficiency of HOTf for the reaction, diazo compound 5a was heated in DCE at 100 °C in the presence of 10 mol % of HOTf. The reaction afforded 3-benzazepine 4a in a yield of 44%, and 51% yield of 5a remained unreacted. As a comparison, we found the reaction delivered 91% yield of 4a when AgOAc (20 mol %) was used as the catalyst, while no reaction occurred without the catalyst.

Therefore, a tentative mechanism for the Ag(I)-catalyzed three-component coupling reaction of 2-alkynylbenzaldehydes, amines, and diazo compounds is proposed in Scheme 5. The condensation of 2-(phenylethynyl)benzaldehyde and aniline affords imine **A**, followed by an intramolecular imine—yne cyclization in the presence of Ag(I) catalyst to give isoquinolinium intermediate **B**.² Intermolecular nucleophilic

Scheme 5. Proposed Reaction Mechanism



attack of diazo compound to B forms intermediate C. The direct 1,2-aryl migration of C leads to the formation of ringexpansion intermediate D with the extrusion of nitrogen (path a). Subsequently, deprotonation and protonation with regeneration of the Ag(I) catalyst produce the 3-benzazepine product. Since AgOAc is also a good catalyst for the conversion of diazo compound 5a to 3-benzazepine and the reaction can be efficiently promoted by increasing the amount of the catalyst loading, another stepwise route was also proposed (path b). The direct deprotonation/protonation from intermediate C gives the isolable diazo compound 5a with concomitant release of AgOAc. The stability of the resulting α -diazo ester 5a is decreased owing to the presence of an amino group on its β position, which could be decomposed by Ag(I) to form the metal-bound carbene intermediate E.^{14b,d} Finally, 1,2-aryl migration through a tricyclic intermediate F gives the ringexpansion product 4a and regenerated the silver catalyst.^{19a,20}

In conclusion, we have developed an efficient threecomponent reaction of 2-alkynylbenzaldehydes, amines, and diazo compounds, leading to 3-benzazepine derivatives by using Ag(I) as the catalyst. The reaction provides a straightforward method for the construction of a biologically important 3benzazepine skeleton from easily available starting materials with the tolerance of a wide range of functional groups. Mechanistically, the reaction is supposed to proceed via a cascade imine—yne cyclization/nucleophilic addition/1,2-aryl migration process. It is the first instance where Ag(I)-mediated 1,2-aryl migration is used in the synthesis of medium-sized heterocycles where no 1,2-H and 1,2-N migration are observed as competitive side reactions.

ASSOCIATED CONTENT

S Supporting Information

This material is available free of charge via the Internet at http://pubs.acs.org. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02159.

Experimental details, characterization data, and NMR spectra of all new products (PDF)

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Notes

The authors declare no competing financial interest.

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